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2,2,2-Trifluoro-1-[3-(2,2,2-trifluoroacetyl)azulen-1-yl]ethanone

Sebastian Förster, Frank Eissmann, Wilhelm Seichter and Edwin Weber*

Institut für Organische Chemie, TU Bergakademie Freiberg, Leipziger Strasse 29, D-09596 Freiberg/Sachsen, Germany
Correspondence e-mail: edwin.weber@chemie.tu-freiberg.de

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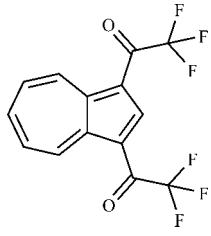
Key indicators: single-crystal X-ray study; T = 100 K; mean σ(C–C) = 0.003 Å; R factor = 0.038; wR factor = 0.106; data-to-parameter ratio = 13.4.

There are two molecules in the asymmetric unit of the title compound, $C_{14}H_6F_6O_2$, in which the azulene systems possess an almost planar geometry with maximum deviations of 0.0438 (15) and 0.0396 (14) Å. Besides intra- and intermolecular C—H···O and C—H···F interactions, the structure displays three F···F contacts [2.793 (2), 2.8820 (17) and 2.9181 (16) Å]. Furthermore, a characteristic azulene π-stacking is observed with an alternating sequence of electron-rich five-membered rings and electron-deficient seven-membered rings [centroid–centroid distances = 3.5413 (12), 3.6847 (12), 3.5790 (12) and 3.7718 (12) Å].

Related literature

For the synthesis, see: Mathias & Overberger (1980); Zielinski *et al.* (2008). For the crystal structure of the parent azulene, see: Robertson *et al.* (1962). For halogen interactions in molecular crystal structures, see: Brammer *et al.* (2001); Metrangolo *et al.* (2008).



Experimental

Crystal data

$C_{14}H_6F_6O_2$
M_r = 320.19
Triclinic, $P\overline{1}$
a = 7.1634 (2) Å
b = 10.8681 (4) Å
c = 16.3286 (5) Å
α = 81.544 (2)°
β = 83.310 (2)°
γ = 80.009 (2)°
V = 1232.92 (7) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 0.18 mm^{-1}
T = 100 K
0.40 × 0.14 × 0.07 mm

Data collection

Bruker APEXII CCD area-detector diffractometer
Absorption correction: multi-scan (*SADABS*; Bruker, 2007)
T_{min} = 0.933, T_{max} = 0.988
20089 measured reflections
5321 independent reflections
3680 reflections with $I > 2\sigma(I)$
R_{int} = 0.031

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.038
$wR(F^2)$ = 0.106
S = 1.02
5321 reflections
397 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.25 e Å^{-3}
$\Delta\rho_{min}$ = −0.31 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
C2—H2···F3	0.95	2.40	2.916 (2)	114
C2A—H2A···F3A	0.95	2.45	2.972 (2)	115
C2A—H2A···F5A	0.95	2.50	2.968 (2)	111
C5—H5···O2	0.95	2.31	2.988 (2)	127
C5A—H5A···O2A	0.95	2.33	3.001 (2)	127
C6—H6···O2A^{i}	0.95	2.52	3.236 (2)	133
C6A—H6A···O2^{i}	0.95	2.47	3.160 (2)	130
C8—H8···F2A^{ii}	0.95	2.45	3.358 (2)	160
C9—H9···O1	0.95	2.31	2.983 (2)	127
C9—H9···O1A^{ii}	0.95	2.58	3.454 (2)	153
C9A—H9A···O1^{ii}	0.95	2.50	3.352 (2)	149
C9A—H9A···O1A	0.95	2.31	2.993 (2)	128

Symmetry codes: (i) $-x+1, -y+2, -z$; (ii) $-x+1, -y+2, -z+1$.

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: IM2284).

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supplementary materials

Acta Cryst. (2011). E**67**, o1419 [**doi:10.1107/S1600536811017569**]

2,2,2-Trifluoro-1-[3-(2,2,2-trifluoroacetyl)azulen-1-yl]ethanone

S. Förster, F. Eissmann, W. Seichter and E. Weber

Comment

The asymmetric unit of the crystal structure contains two molecules featuring planar azulene ring systems (Fig 1). In the crystallographic *b* and *c* directions the structure is stabilized by hydrogen bonding both between the carbonyl oxygen or fluorine atoms and hydrogen atoms of the electronic deficient seven-membered ring (Fig 2). The crystal structure is characterized by formation of molecular stacks along the crystallographic *a*-axis (Fig 3). Due to the inherent dipole character of the azulene ring system, the molecules are arranged in a head-to-tail fashion within the stacks. Each molecule in the asymmetric unit shows two different interactions to its stack neighbors, which leads to four different distances between the centroids *Cg*1···*Cg*2^{i}, *d* = 3.5413 (12), *Cg*1···*Cg*2^{ii}, *d* = 3.6847 (12), *Cg*1A···*Cg*2A^{iii}, *d* = 3.5790 (12), *Cg*1A···*Cg*2A^{iv}, *d* = 3.7718 (12). Furthermore, three halogen···halogen contacts were observed: F5···F5 [*d* = 2.7931 (23) Å, $\theta_1 = \theta_2$ = 145.79 (14)°], F3···F6A [*d* = 2.8820 (17), θ_1 = 124.02 (11), θ_2 = 113.52 (12)] and F6A···F6 [*d* = 2.9181 (16), θ_1 = 161.89 (11), θ_2 = 156.90 (12)]. Symmetry codes: (i) = -*x*, 2 - *y*, -*z*; (ii) = 1 - *x*, 2 - *y*, -*z*; (iii) 1 - *x*, 2 - *y*, 1 - *z*; (iv) 2 - *x*, 2 - *y*, 1 - *z*.

Experimental

The title compound was prepared according to the literature procedure of Mathias *et al.* (1980). Crystallization by slow evaporation from acetone yielded suitable crystals after 3 days.

Refinement

Aromatic H atoms were positioned geometrically and allowed to ride on their respective parent atoms, with C—H = 0.95 Å and U_{iso} = 1.2 U_{eq}(C).

Figures

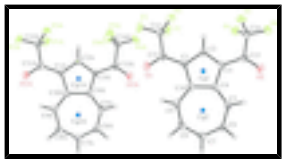


Fig. 1. Asymmetric unit of the title compound, showing the atom numbering schemes. Displacement ellipsoids are drawn at the 50% probability level.

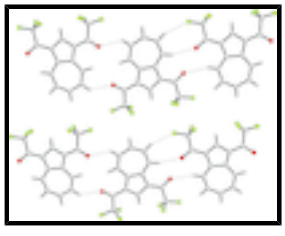


Fig. 2. Molecular sheets within the packing of the title compound. Intermolecular interactions are represented as dashed lines.

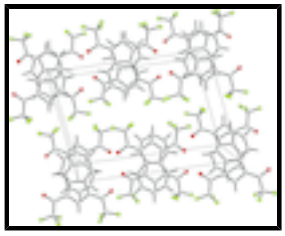


Fig. 3. Packing diagram viewed down the *a* axis, showing the stacking interactions of the azulene systems.

2,2,2-Trifluoro-1-[3-(2,2,2-trifluoroacetyl)azulen-1-yl]ethanone

Crystal data

$C_{14}H_6F_6O_2$	$Z = 4$
$M_r = 320.19$	$F(000) = 640$
Triclinic, $P\bar{1}$	$D_x = 1.725$ Mg m^{-3}
Hall symbol: -P 1	Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
$a = 7.1634$ (2) Å	Cell parameters from 6112 reflections
$b = 10.8681$ (4) Å	$\theta = 2.9$–$28.9°$
$c = 16.3286$ (5) Å	$\mu = 0.18$ mm^{-1}
$\alpha = 81.544$ (2)°	$T = 100$ K
$\beta = 83.310$ (2)°	Rod, red
$\gamma = 80.009$ (2)°	0.40 × 0.14 × 0.07 mm
$V = 1232.92$ (7) Å^3	

Data collection

Bruker APEXII CCD area-detector diffractometer	5321 independent reflections
Radiation source: fine-focus sealed tube	3680 reflections with $I > 2\sigma(I)$
graphite	$R_{int} = 0.031$
φ and ω scans	$\theta_{max} = 27.1°$, $\theta_{min} = 1.3°$
Absorption correction: multi-scan (*SADABS*; Bruker, 2007)	$h = -9 \rightarrow 9$
$T_{min} = 0.933$, $T_{max} = 0.988$	$k = -13 \rightarrow 13$
20089 measured reflections	$l = -20 \rightarrow 20$

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.038$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.106$	H-atom parameters constrained
$S = 1.02$	$w = 1/[\sigma^2(F_o^2) + (0.0465P)^2 + 0.5715P]$ where $P = (F_o^2 + 2F_c^2)/3$
5321 reflections	$(\Delta/\sigma)_{max} < 0.001$
397 parameters	$\Delta\rho_{max} = 0.25$ e Å^{-3}

0 restraints $\Delta\rho_{min} = -0.31$ e Å^{-3}

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2, conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
F1	0.39463 (18)	0.60356 (13)	0.19502 (8)	0.0398 (3)
F2	0.2168 (2)	0.64770 (12)	0.30613 (7)	0.0412 (3)
F3	0.09017 (18)	0.61716 (11)	0.19949 (7)	0.0321 (3)
F4	0.36566 (18)	0.57475 (11)	−0.07337 (8)	0.0346 (3)
F5	0.05955 (18)	0.59590 (11)	−0.05863 (7)	0.0344 (3)
F6	0.2086 (2)	0.60122 (12)	−0.18071 (7)	0.0370 (3)
O1	0.2103 (2)	0.88041 (15)	0.24264 (9)	0.0379 (4)
O2	0.1713 (2)	0.84185 (14)	−0.17976 (8)	0.0319 (4)
C1	0.2215 (3)	0.84447 (19)	0.10164 (11)	0.0204 (4)
C2	0.2071 (3)	0.76448 (18)	0.04375 (11)	0.0195 (4)
H2	0.1961	0.6778	0.0568	0.023*
C3	0.2114 (3)	0.83168 (18)	−0.03648 (11)	0.0192 (4)
C4	0.2344 (2)	0.95792 (18)	−0.03047 (11)	0.0188 (4)
C5	0.2515 (3)	1.05229 (18)	−0.09710 (12)	0.0213 (4)
H5	0.2413	1.0298	−0.1503	0.026*
C6	0.2813 (3)	1.17487 (19)	−0.09636 (12)	0.0237 (4)
H6	0.2861	1.2254	−0.1490	0.028*
C7	0.3050 (3)	1.23385 (19)	−0.02918 (13)	0.0246 (4)
H7	0.3251	1.3189	−0.0424	0.030*
C8	0.3033 (3)	1.18607 (19)	0.05439 (13)	0.0240 (4)
H8	0.3249	1.2427	0.0901	0.029*
C9	0.2749 (3)	1.06782 (19)	0.09354 (12)	0.0223 (4)
H9	0.2777	1.0549	0.1522	0.027*
C10	0.2428 (3)	0.96584 (18)	0.05790 (11)	0.0194 (4)
C11	0.2178 (3)	0.8089 (2)	0.19135 (12)	0.0243 (4)
C12	0.2291 (3)	0.6679 (2)	0.22329 (12)	0.0280 (5)
C13	0.1949 (3)	0.78107 (19)	−0.11202 (12)	0.0223 (4)
C14	0.2074 (3)	0.6369 (2)	−0.10652 (12)	0.0257 (5)
F1A	0.82283 (17)	0.62221 (12)	0.68301 (8)	0.0367 (3)
F2A	0.6196 (2)	0.67680 (12)	0.78398 (7)	0.0382 (3)
F3A	0.52235 (17)	0.63290 (11)	0.67388 (7)	0.0307 (3)

F4A	0.86640 (18)	0.55033 (11)	0.42077 (8)	0.0373 (3)
F5A	0.56351 (17)	0.59936 (12)	0.41519 (8)	0.0346 (3)
F6A	0.7480 (2)	0.57401 (12)	0.30349 (8)	0.0413 (3)
O1A	0.6058 (2)	0.90591 (14)	0.71153 (8)	0.0314 (4)
O2A	0.7788 (2)	0.80970 (14)	0.29118 (8)	0.0320 (4)
C1A	0.6844 (3)	0.85148 (18)	0.57393 (11)	0.0185 (4)
C2A	0.6948 (3)	0.76475 (18)	0.51849 (11)	0.0193 (4)
H2A	0.6756	0.6797	0.5335	0.023*
C3A	0.7379 (3)	0.82104 (18)	0.43672 (11)	0.0187 (4)
C4A	0.7618 (2)	0.94807 (18)	0.44046 (11)	0.0175 (4)
C5A	0.8140 (3)	1.03354 (18)	0.37289 (12)	0.0203 (4)
H5A	0.8301	1.0040	0.3202	0.024*
C6A	0.8452 (3)	1.15565 (18)	0.37253 (12)	0.0217 (4)
H6A	0.8807	1.1984	0.3194	0.026*
C7A	0.8324 (3)	1.22506 (18)	0.43871 (12)	0.0218 (4)
H7A	0.8577	1.3089	0.4243	0.026*
C8A	0.7876 (3)	1.18895 (18)	0.52299 (12)	0.0212 (4)
H8A	0.7898	1.2507	0.5583	0.025*
C9A	0.7401 (3)	1.07489 (18)	0.56275 (12)	0.0203 (4)
H9A	0.7133	1.0694	0.6214	0.024*
C10A	0.7267 (2)	0.96792 (17)	0.52800 (11)	0.0174 (4)
C11A	0.6426 (3)	0.82740 (19)	0.66380 (12)	0.0220 (4)
C12A	0.6515 (3)	0.6882 (2)	0.70151 (12)	0.0254 (4)
C13A	0.7520 (3)	0.76079 (19)	0.36250 (12)	0.0223 (4)
C14A	0.7319 (3)	0.6197 (2)	0.37514 (13)	0.0271 (5)

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
F1	0.0358 (7)	0.0393 (8)	0.0377 (7)	0.0061 (6)	−0.0025 (6)	0.0019 (6)
F2	0.0649 (9)	0.0404 (8)	0.0172 (6)	−0.0087 (7)	−0.0069 (6)	0.0030 (5)
F3	0.0401 (7)	0.0298 (7)	0.0280 (7)	−0.0146 (6)	−0.0019 (5)	0.0006 (5)
F4	0.0390 (7)	0.0265 (7)	0.0354 (7)	0.0019 (6)	−0.0017 (6)	−0.0048 (5)
F5	0.0405 (8)	0.0299 (7)	0.0340 (7)	−0.0169 (6)	0.0104 (6)	−0.0059 (5)
F6	0.0578 (9)	0.0325 (7)	0.0242 (7)	−0.0138 (6)	0.0020 (6)	−0.0121 (5)
O1	0.0591 (11)	0.0384 (9)	0.0216 (8)	−0.0204 (8)	−0.0042 (7)	−0.0064 (7)
O2	0.0502 (10)	0.0284 (8)	0.0186 (8)	−0.0108 (7)	−0.0062 (7)	−0.0003 (6)
C1	0.0179 (9)	0.0263 (11)	0.0178 (10)	−0.0050 (8)	−0.0018 (7)	−0.0034 (8)
C2	0.0168 (9)	0.0215 (10)	0.0199 (10)	−0.0039 (8)	−0.0011 (7)	−0.0015 (8)
C3	0.0172 (9)	0.0225 (10)	0.0182 (9)	−0.0039 (8)	−0.0009 (7)	−0.0033 (7)
C4	0.0131 (9)	0.0228 (10)	0.0205 (10)	−0.0030 (7)	−0.0013 (7)	−0.0031 (8)
C5	0.0174 (10)	0.0264 (11)	0.0206 (10)	−0.0036 (8)	−0.0029 (7)	−0.0044 (8)
C6	0.0213 (10)	0.0245 (11)	0.0248 (11)	−0.0054 (8)	−0.0031 (8)	0.0011 (8)
C7	0.0198 (10)	0.0203 (11)	0.0342 (12)	−0.0040 (8)	−0.0026 (8)	−0.0038 (8)
C8	0.0199 (10)	0.0235 (11)	0.0310 (11)	−0.0040 (8)	−0.0048 (8)	−0.0086 (8)
C9	0.0168 (9)	0.0285 (11)	0.0228 (10)	−0.0029 (8)	−0.0052 (8)	−0.0055 (8)
C10	0.0130 (9)	0.0250 (11)	0.0206 (10)	−0.0035 (8)	−0.0015 (7)	−0.0032 (8)
C11	0.0239 (10)	0.0310 (12)	0.0196 (10)	−0.0082 (9)	−0.0030 (8)	−0.0033 (8)

C12	0.0336 (12)	0.0314 (12)	0.0179 (10)	−0.0046 (9)	−0.0029 (8)	0.0000 (8)
C13	0.0216 (10)	0.0261 (11)	0.0195 (10)	−0.0061 (8)	0.0017 (8)	−0.0044 (8)
C14	0.0318 (11)	0.0277 (11)	0.0182 (10)	−0.0078 (9)	0.0034 (8)	−0.0055 (8)
F1A	0.0294 (7)	0.0335 (7)	0.0415 (8)	−0.0003 (6)	−0.0045 (6)	0.0092 (6)
F2A	0.0566 (9)	0.0413 (8)	0.0189 (6)	−0.0200 (7)	−0.0048 (6)	0.0043 (5)
F3A	0.0363 (7)	0.0295 (7)	0.0293 (7)	−0.0159 (5)	−0.0057 (5)	0.0010 (5)
F4A	0.0389 (7)	0.0231 (7)	0.0484 (8)	−0.0008 (6)	−0.0043 (6)	−0.0048 (6)
F5A	0.0330 (7)	0.0371 (8)	0.0383 (7)	−0.0175 (6)	0.0070 (5)	−0.0142 (6)
F6A	0.0581 (9)	0.0388 (8)	0.0341 (7)	−0.0221 (7)	0.0097 (6)	−0.0217 (6)
O1A	0.0429 (9)	0.0332 (9)	0.0212 (7)	−0.0150 (7)	0.0010 (6)	−0.0065 (6)
O2A	0.0481 (10)	0.0293 (8)	0.0194 (8)	−0.0081 (7)	−0.0021 (7)	−0.0043 (6)
C1A	0.0158 (9)	0.0224 (10)	0.0180 (9)	−0.0038 (8)	−0.0034 (7)	−0.0027 (7)
C2A	0.0156 (9)	0.0191 (10)	0.0229 (10)	−0.0030 (7)	−0.0038 (7)	−0.0001 (8)
C3A	0.0167 (9)	0.0219 (10)	0.0184 (9)	−0.0039 (8)	−0.0022 (7)	−0.0043 (8)
C4A	0.0130 (9)	0.0209 (10)	0.0181 (9)	−0.0014 (7)	−0.0034 (7)	−0.0011 (7)
C5A	0.0177 (9)	0.0245 (11)	0.0185 (9)	−0.0017 (8)	−0.0023 (7)	−0.0042 (8)
C6A	0.0190 (10)	0.0247 (11)	0.0204 (10)	−0.0042 (8)	−0.0016 (8)	0.0012 (8)
C7A	0.0176 (10)	0.0207 (10)	0.0270 (11)	−0.0034 (8)	−0.0047 (8)	−0.0009 (8)
C8A	0.0179 (10)	0.0201 (10)	0.0267 (11)	−0.0024 (8)	−0.0043 (8)	−0.0060 (8)
C9A	0.0150 (9)	0.0266 (11)	0.0195 (10)	−0.0018 (8)	−0.0036 (7)	−0.0043 (8)
C10A	0.0118 (9)	0.0220 (10)	0.0186 (9)	−0.0025 (7)	−0.0021 (7)	−0.0026 (7)
C11A	0.0186 (10)	0.0297 (11)	0.0193 (10)	−0.0082 (8)	−0.0026 (7)	−0.0024 (8)
C12A	0.0286 (11)	0.0308 (12)	0.0179 (10)	−0.0098 (9)	−0.0031 (8)	0.0004 (8)
C13A	0.0203 (10)	0.0262 (11)	0.0217 (10)	−0.0051 (8)	−0.0022 (8)	−0.0054 (8)
C14A	0.0298 (11)	0.0284 (12)	0.0256 (11)	−0.0079 (9)	0.0021 (9)	−0.0119 (9)

Geometric parameters (Å, °)

F1—C12	1.337 (2)	F1A—C12A	1.336 (2)
F2—C12	1.333 (2)	F2A—C12A	1.330 (2)
F3—C12	1.337 (2)	F3A—C12A	1.338 (2)
F4—C14	1.343 (2)	F4A—C14A	1.342 (3)
F5—C14	1.341 (2)	F5A—C14A	1.341 (2)
F6—C14	1.324 (2)	F6A—C14A	1.322 (2)
O1—C11	1.214 (2)	O1A—C11A	1.214 (2)
O2—C13	1.216 (2)	O2A—C13A	1.215 (2)
C1—C2	1.398 (3)	C1A—C2A	1.387 (3)
C1—C10	1.428 (3)	C1A—C10A	1.434 (3)
C1—C11	1.457 (3)	C1A—C11A	1.457 (3)
C2—C3	1.403 (3)	C2A—C3A	1.408 (3)
C2—H2	0.9500	C2A—H2A	0.9500
C3—C4	1.429 (3)	C3A—C4A	1.431 (3)
C3—C13	1.447 (3)	C3A—C13A	1.446 (3)
C4—C5	1.391 (3)	C4A—C5A	1.394 (3)
C4—C10	1.467 (3)	C4A—C10A	1.463 (2)
C5—C6	1.387 (3)	C5A—C6A	1.383 (3)
C5—H5	0.9500	C5A—H5A	0.9500
C6—C7	1.390 (3)	C6A—C7A	1.392 (3)
C6—H6	0.9500	C6A—H6A	0.9500

C7—C8	1.386 (3)	C7A—C8A	1.390 (3)
C7—H7	0.9500	C7A—H7A	0.9500
C8—C9	1.386 (3)	C8A—C9A	1.390 (3)
C8—H8	0.9500	C8A—H8A	0.9500
C9—C10	1.389 (3)	C9A—C10A	1.389 (3)
C9—H9	0.9500	C9A—H9A	0.9500
C11—C12	1.536 (3)	C11A—C12A	1.541 (3)
C13—C14	1.544 (3)	C13A—C14A	1.546 (3)
C2—C1—C10	108.52 (16)	C2A—C1A—C10A	108.39 (16)
C2—C1—C11	125.49 (18)	C2A—C1A—C11A	125.81 (17)
C10—C1—C11	125.99 (17)	C10A—C1A—C11A	125.78 (17)
C1—C2—C3	109.69 (17)	C1A—C2A—C3A	110.12 (17)
C1—C2—H2	125.2	C1A—C2A—H2A	124.9
C3—C2—H2	125.2	C3A—C2A—H2A	124.9
C2—C3—C4	108.25 (16)	C2A—C3A—C4A	107.87 (16)
C2—C3—C13	125.54 (18)	C2A—C3A—C13A	125.59 (18)
C4—C3—C13	126.20 (17)	C4A—C3A—C13A	126.53 (17)
C5—C4—C3	125.52 (17)	C5A—C4A—C3A	125.60 (17)
C5—C4—C10	127.61 (17)	C5A—C4A—C10A	127.54 (17)
C3—C4—C10	106.84 (16)	C3A—C4A—C10A	106.85 (15)
C6—C5—C4	128.73 (18)	C6A—C5A—C4A	128.50 (18)
C6—C5—H5	115.6	C6A—C5A—H5A	115.7
C4—C5—H5	115.6	C4A—C5A—H5A	115.7
C5—C6—C7	128.88 (19)	C5A—C6A—C7A	129.35 (18)
C5—C6—H6	115.6	C5A—C6A—H6A	115.3
C7—C6—H6	115.6	C7A—C6A—H6A	115.3
C8—C7—C6	129.13 (19)	C8A—C7A—C6A	129.10 (19)
C8—C7—H7	115.4	C8A—C7A—H7A	115.5
C6—C7—H7	115.4	C6A—C7A—H7A	115.5
C7—C8—C9	129.54 (19)	C7A—C8A—C9A	128.92 (18)
C7—C8—H8	115.2	C7A—C8A—H8A	115.5
C9—C8—H8	115.2	C9A—C8A—H8A	115.5
C10—C9—C8	128.27 (18)	C10A—C9A—C8A	128.53 (18)
C10—C9—H9	115.9	C10A—C9A—H9A	115.7
C8—C9—H9	115.9	C8A—C9A—H9A	115.7
C9—C10—C1	125.46 (17)	C9A—C10A—C1A	125.14 (17)
C9—C10—C4	127.81 (18)	C9A—C10A—C4A	128.04 (17)
C1—C10—C4	106.65 (16)	C1A—C10A—C4A	106.74 (15)
O1—C11—C1	125.8 (2)	O1A—C11A—C1A	126.36 (19)
O1—C11—C12	117.38 (18)	O1A—C11A—C12A	117.31 (17)
C1—C11—C12	116.78 (17)	C1A—C11A—C12A	116.31 (17)
F2—C12—F1	107.72 (17)	F2A—C12A—F1A	107.42 (16)
F2—C12—F3	107.00 (17)	F2A—C12A—F3A	107.06 (16)
F1—C12—F3	107.28 (17)	F1A—C12A—F3A	107.55 (16)
F2—C12—C11	111.20 (17)	F2A—C12A—C11A	111.26 (16)
F1—C12—C11	110.51 (17)	F1A—C12A—C11A	110.94 (16)
F3—C12—C11	112.90 (17)	F3A—C12A—C11A	112.37 (16)
O2—C13—C3	125.94 (19)	O2A—C13A—C3A	126.57 (19)
O2—C13—C14	116.69 (17)	O2A—C13A—C14A	116.71 (17)

C3—C13—C14	117.37 (17)	C3A—C13A—C14A	116.72 (17)
F6—C14—F5	107.30 (16)	F6A—C14A—F4A	107.38 (17)
F6—C14—F4	107.41 (16)	F6A—C14A—F5A	107.43 (16)
F5—C14—F4	106.68 (16)	F4A—C14A—F5A	106.64 (17)
F6—C14—C13	111.59 (16)	F6A—C14A—C13A	111.71 (17)
F5—C14—C13	111.68 (16)	F4A—C14A—C13A	111.30 (16)
F4—C14—C13	111.89 (16)	F5A—C14A—C13A	112.10 (16)
C10—C1—C2—C3	−2.2 (2)	C10A—C1A—C2A—C3A	−1.5 (2)
C11—C1—C2—C3	178.56 (18)	C11A—C1A—C2A—C3A	179.96 (17)
C1—C2—C3—C4	1.6 (2)	C1A—C2A—C3A—C4A	1.8 (2)
C1—C2—C3—C13	−178.33 (18)	C1A—C2A—C3A—C13A	−177.22 (17)
C2—C3—C4—C5	177.72 (17)	C2A—C3A—C4A—C5A	177.00 (18)
C13—C3—C4—C5	−2.3 (3)	C13A—C3A—C4A—C5A	−4.0 (3)
C2—C3—C4—C10	−0.4 (2)	C2A—C3A—C4A—C10A	−1.4 (2)
C13—C3—C4—C10	179.51 (18)	C13A—C3A—C4A—C10A	177.65 (18)
C3—C4—C5—C6	−177.32 (19)	C3A—C4A—C5A—C6A	−177.74 (18)
C10—C4—C5—C6	0.4 (3)	C10A—C4A—C5A—C6A	0.3 (3)
C4—C5—C6—C7	1.3 (3)	C4A—C5A—C6A—C7A	−0.3 (3)
C5—C6—C7—C8	−0.6 (4)	C5A—C6A—C7A—C8A	1.2 (3)
C6—C7—C8—C9	−1.1 (4)	C6A—C7A—C8A—C9A	−1.5 (3)
C7—C8—C9—C10	0.9 (4)	C7A—C8A—C9A—C10A	0.4 (3)
C8—C9—C10—C1	177.52 (19)	C8A—C9A—C10A—C1A	177.21 (18)
C8—C9—C10—C4	1.1 (3)	C8A—C9A—C10A—C4A	0.9 (3)
C2—C1—C10—C9	−175.17 (18)	C2A—C1A—C10A—C9A	−176.37 (17)
C11—C1—C10—C9	4.1 (3)	C11A—C1A—C10A—C9A	2.2 (3)
C2—C1—C10—C4	1.8 (2)	C2A—C1A—C10A—C4A	0.6 (2)
C11—C1—C10—C4	−178.89 (18)	C11A—C1A—C10A—C4A	179.14 (17)
C5—C4—C10—C9	−2.0 (3)	C5A—C4A—C10A—C9A	−1.0 (3)
C3—C4—C10—C9	176.06 (18)	C3A—C4A—C10A—C9A	177.34 (18)
C5—C4—C10—C1	−178.95 (18)	C5A—C4A—C10A—C1A	−177.85 (18)
C3—C4—C10—C1	−0.9 (2)	C3A—C4A—C10A—C1A	0.47 (19)
C2—C1—C11—O1	−171.8 (2)	C2A—C1A—C11A—O1A	−168.50 (19)
C10—C1—C11—O1	9.0 (3)	C10A—C1A—C11A—O1A	13.2 (3)
C2—C1—C11—C12	9.7 (3)	C2A—C1A—C11A—C12A	13.2 (3)
C10—C1—C11—C12	−169.47 (18)	C10A—C1A—C11A—C12A	−165.03 (17)
O1—C11—C12—F2	3.9 (3)	O1A—C11A—C12A—F2A	−1.3 (3)
C1—C11—C12—F2	−177.52 (17)	C1A—C11A—C12A—F2A	177.14 (16)
O1—C11—C12—F1	−115.7 (2)	O1A—C11A—C12A—F1A	−120.81 (19)
C1—C11—C12—F1	62.9 (2)	C1A—C11A—C12A—F1A	57.6 (2)
O1—C11—C12—F3	124.1 (2)	O1A—C11A—C12A—F3A	118.75 (19)
C1—C11—C12—F3	−57.2 (2)	C1A—C11A—C12A—F3A	−62.8 (2)
C2—C3—C13—O2	168.8 (2)	C2A—C3A—C13A—O2A	175.72 (19)
C4—C3—C13—O2	−11.1 (3)	C4A—C3A—C13A—O2A	−3.1 (3)
C2—C3—C13—C14	−10.9 (3)	C2A—C3A—C13A—C14A	−4.6 (3)
C4—C3—C13—C14	169.21 (17)	C4A—C3A—C13A—C14A	176.51 (17)
O2—C13—C14—F6	6.2 (3)	O2A—C13A—C14A—F6A	0.0 (3)
C3—C13—C14—F6	−174.04 (16)	C3A—C13A—C14A—F6A	−179.65 (17)
O2—C13—C14—F5	−113.9 (2)	O2A—C13A—C14A—F4A	120.0 (2)
C3—C13—C14—F5	65.9 (2)	C3A—C13A—C14A—F4A	−59.6 (2)

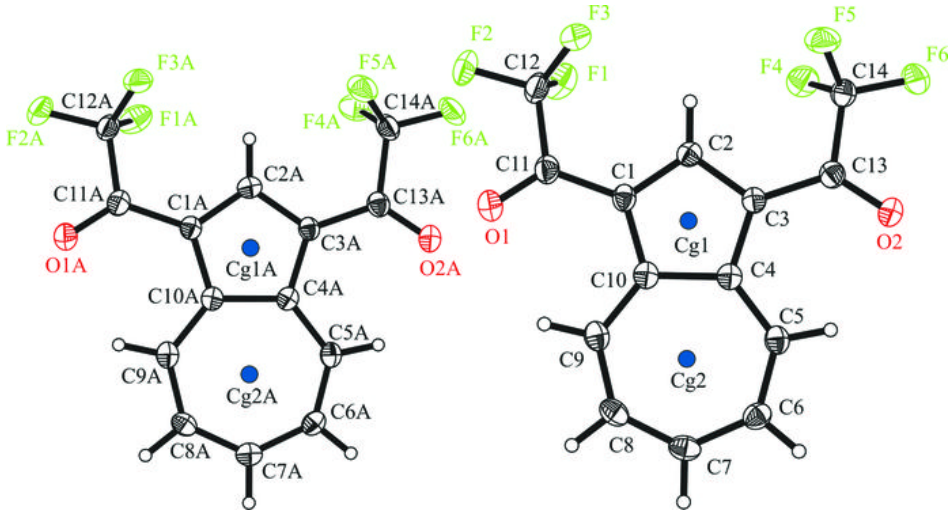
O2—C13—C14—F4	126.63 (19)	O2A—C13A—C14A—F5A	−120.6 (2)
C3—C13—C14—F4	−53.7 (2)	C3A—C13A—C14A—F5A	59.7 (2)

Hydrogen-bond geometry (Å, °)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
C2—H2···F3	0.95	2.40	2.916 (2)	114
C2A—H2A···F3A	0.95	2.45	2.972 (2)	115
C2A—H2A···F5A	0.95	2.50	2.968 (2)	111
C5—H5···O2	0.95	2.31	2.988 (2)	127
C5A—H5A···O2A	0.95	2.33	3.001 (2)	127
C6—H6···O2A^{i}	0.95	2.52	3.236 (2)	133
C6A—H6A···O2^{i}	0.95	2.47	3.160 (2)	130
C8—H8···F2A^{ii}	0.95	2.45	3.358 (2)	160
C9—H9···O1	0.95	2.31	2.983 (2)	127
C9—H9···O1A^{ii}	0.95	2.58	3.454 (2)	153
C9A—H9A···O1^{ii}	0.95	2.50	3.352 (2)	149
C9A—H9A···O1A	0.95	2.31	2.993 (2)	128

Symmetry codes: (i) −*x*+1, −*y*+2, −*z*; (ii) −*x*+1, −*y*+2, −*z*+1.

Fig. 1


F3A
C12A
F2A
F1A
C11A
O1A
C1A
C2A
C13A
F5A
F4A
C14A
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C3A
O2A
Cg1A
C10A
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C5A
C9A
Cg2A
C6A
C8A
C7A
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F3
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F1
C11
O1
C1
C2
C13
F5
F4
C14
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O2
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C10
C4
C5
C9
Cg2
C6
C8
C7

Fig. 2

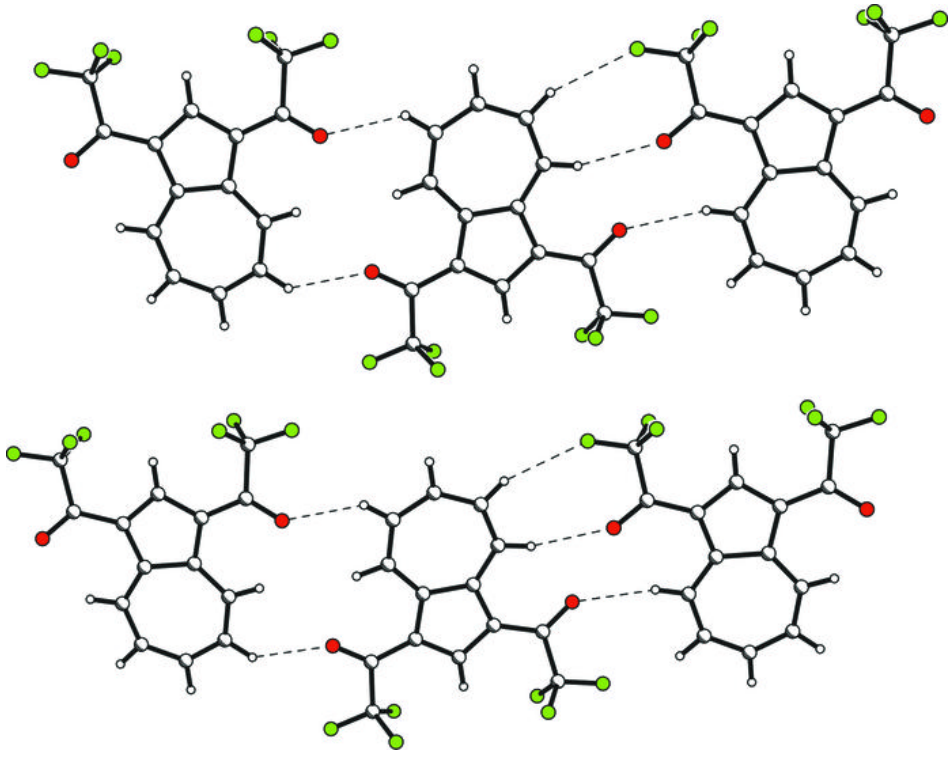

Fig. 3